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Synthesis and biological evaluation of 12 allenic aromatic ethers

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Abstract—Twelve allenic aromatic ethers, some of them are natural products isolated from the mangrove fungus *Xylaria* sp. 2508 in the South China Sea, were synthesized. Their antitumor activities against KB and KBv200 cells were determined. All these compounds demonstrated cytotoxic potential, ranging from weak to strong activity. The analysis of structure–activity relationships suggested that the introduction of allenic moiety could generate or enhance cytotoxicity of these phenol compounds. © 2007 Elsevier Ltd. All rights reserved.

About 150 natural products with an allenic or cumulenic structure moiety are known today. Inspired by the intriguing biological activities of many allenic natural products, allenic moieties are now introduced in the compounds which have the pharmacologically activity. The many functionalized allenes thus obtained exhibit impressive activities, such as enzyme inhibitor activities, cytotoxic or antiviral activities, etc. ¹

Naturally occurring allenic compounds can be found in microorganisms, fungi, higher plants, and insects. So far, only several allenic aromatic ethers, for example, chestersiene from the Hypoxylon chestersii,2 2b from the Clitocybe eucalyptorum, xyloallenoide A and 2f from the mangrove fungus Xylaria sp. 2508⁴ were isolated. Recently, we isolated four new allenic aromatic ethers (2c, **3a–c**) from the *Xylaria* sp. 2508 again. ⁵ In the preliminary bioassay, some natural allenic aromatic ethers showed antitumor activities against KB and KBv200 cells. The limited amounts of isolated materials prevented the study on their biological activities. Therefore, we synthesized 12 allenic aromatic ether analogs (see Fig. 1) to further explore the effect of allenic groups bioactivities of compounds. These compounds are the derivatives of allenic group linking to compounds 1. Compound 2i is analogous to the natural product xyloallenoide A.

 $\it Keywords$: Allenic aromatic ethers; Synthesis; Antitumor agent; KB cell; KBv200 cell.

In the synthesis of these allenic aromatic ethers, one key step was the preparation of 2,3-butadiene-1-ol (6). There were two methods to synthesize 6 (Scheme 1). The first method used 2-butyne-1,4-diol (4) as the starting material. Chlorination of 4 with thionyl chloride gave the monochloroalkyne (5), which was reduced with lithium aluminum hydride to give compound 6,6 but in our work, the yield of 6 was very low (15%), it was difficult to separate the monochloroalkyne (5) from the dichloroalkyne, and compound 5 was a severe skin irritant. We tried another method that Baeckstrom et al. reported,7 which started through a Mannich reaction with diethylamine, formaldehyde, and 2-propynol. The resulting 4-diethylamine-2-butyn-1-ol (7) was methylated with dimethyl sulfate, then reduced with lithium hydride to afford 6 with a high yield of 71%. So the latter method was used to synthesize a series of compound 2, and 3.

Because 1a-h were commercially available and 6 was also prepared, thus the series of compound 2 could be synthesized readily. Scheme 2 shows the synthesis of the compounds 2a-j. 4-Bromo-1,2-butadiene (8) was prepared by bromination of 6 with phosphorus tribromide and trace amount of pyridine at 0 °C.8

Then, treatment of compound 8 with 1a-d and 1g-j in the presence of potassium carbonate gave 2a-d and 2g-j, respectively. Hydrolysis of compounds 2a and 2b with LiOH in THF-H₂O provided the corresponding acids 2e and 2f.

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Figure 1. Structures of compounds 1a-j, 2a-j, and 3a-c.

(1) HO
$$\frac{\text{SOCl}_2}{\text{CI}}$$
 HO $\frac{\text{LiAIH}_4}{\text{5}}$ =C OH

(2) OH + HCHO + $(C_2H_5)_2NH$ $\frac{\text{CuSO}_4}{\text{7}}$ OH $\frac{1)(CH_3)_2SO_4}{2)\text{LiAIH}_4}$ 6

Scheme 1. Two methods to synthesize 6.

$$= C = \underbrace{\begin{array}{c} PBr_3 \\ 6 \end{array}}_{OH} \underbrace{\begin{array}{c} PBr_3 \\ Pyridine \end{array}}_{8} = C = \underbrace{\begin{array}{c} R^1 \\ K_2CO_3/DMF \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2 \end{array}}_{2} = C = \underbrace{\begin{array}{c} R^1 \\ C \\ 2$$

Scheme 2. Synthesis of 2a-j.

In an attempt to study compound 2 analogs with two allenic moieties, we used an excess amount of compound 8 to react with caffeic acid methyl ester (1d) or chlorogenic acid (1g). Because the molecules of 1d and 1g contain two phenol hydroxyl groups, it was anticipated that they would form two allenic ethers in the excess amount of 8 (mole ratio 2.5:1 or 3.5:1). However, it was unexpected that only one OH formed allenic ether, the OH of R¹ did not react with 8 in the two compounds. Moreover, the carboxyl group of 1g reacted with one allenic moiety formed the ester. The structures of these compounds were determined by the spectral data. Furthermore, the single crystals of 2g were obtained by crystallization in EtOAc and PE. The structure of 2g was finally confirmed by X-ray diffractive technique.

The method described in Scheme 3 allowed for the preparation of 1j. The intermediate 1-amino-cyclo-propane-carboxylic acid methyl ester (10) was synthesized via

1-amino-cyclopropanecarboxylic (9) in the presence of thionyl chloride in methanol. Compound 10 then coupled with compound 1f in the presence of BOP and DIEA in DMF to afford 1j.

The synthesis of the compounds **3a–c** is presented in Scheme **4**. Several methods for preparation of halide of **2h** were examined. When **2h** was treated with phosphorus tribromide or thionyl chloride, only a small amount of the corresponding halide was obtained. We tried to use the Ph₃P/I₂ reagent. The iodide compound **11** was successfully synthesized with a good yield of 82%. Treatment of **11** with **1a–c** in the presence of potassium carbonate finally gave compounds **3a–c**, respectively.

The 12 allenic aromatic ethers, together with 1a–j, 6, 8, and 10, were evaluated in vitro for their ability to inhibit the growth of short-term cultured tongue cancer KB and KBv200 cell lines by a modified MTT (3-(4,5-dimethyl-

Scheme 3. Synthesis of 1j.

Scheme 4. Synthesis of 3a-c.

Table 1. Inhibitory activities of phenols and allenic aromatic ethers against tongue cancer KB and KBv200 cells in vitro

No.	$IC_{50}^{a} (\mu M)$		No.	$IC_{50}^{a}(\mu M)$		No.	$IC_{50}^{a} (\mu M)$	
	KB	KBv200		KB	KBv200		KB	KBv200
1a	>200	>200	2a	22.52	20.86	3a	75.40	52.92
1b	123.1	101.2	2b	27.64	18.96	3b	48.76	55.47
1c	>200	>200	2c	17.36	18.27	3c	34.55	30.27
1d	>200	>200	2d	29.36	23.38	6	>200	>200
1e	>200	>200	2e	40.58	49.56	8	>200	>200
1f	>200	>200	2 f	71.5	52.42	10	135.2	>200
1g	>200	>200	2g	53.10	35.68			
1h	>200	>200	2h	74.45	76.58			
1j	>200	>200	2j	48.06	77.03			

^a Each experiment was independently performed three times and expressed as means. In addition, variation between the replicates was not greater than 4% of any given value.

thiazol-2-yl)-2,5-diphenyl tetrazolium bromide) assay. 11,12 The IC $_{50}$ values were also determined (Table 1).

As shown in Table 1, phenols, such as 1a and 1c-i, have no inhibitory activities against KB or KBv200 cells. while the corresponding allenic compounds 2a and 2ci exhibited inhibitory activities against these two cells. Compound 1b itself has weak inhibitory activity, with the IC₅₀ values for KB/KBv200 of 123.1/101.2 μ M, but the corresponding allenic compound 2b has stronger activity, with IC₅₀ values for KB/KBv200 of 27.64/ 18.96 µM. In fact, all the 12 allenic aromatic ethers exhibited cytotoxic potential, ranging from weak to strong activity. These data indicated that the introduction of allenic moiety to the phenol hydroxyl group could generate or enhance cytotoxicity of these phenols. Although the experiments were preliminary, it suggests that the allenic moiety is effective for the antitumor activities of these compounds.

Among the 12 allenic aromatic ethers, four compounds ${\bf 2a-d}$ have significant cytotoxicities (IC₅₀ < 30 μ M). The introduction of OCH₃ (2c) or OH (2d) groups on the benzene ring of 2b showed no influence on the activity. Compound 2g has two allenic moieties (one is linked to the carboxyl group), but it did not exhibit better activity. Although compound 10 has weak inhibitory activity against KB cells with IC₅₀ value of 135.2 μ M,

the introduction of this moiety into 2b to form compound 2j did not obtain the desired activity. It was also observed that simple allenic compounds 6 and 8 had no inhibitory activities against KB or KBv200 cells, whereas the allenic aromatic ethers 2 or 3 showed good inhibitory activities. Interestingly, when 2 and 3 have the same R^1 , R^2 group, 2 showed better inhibitory activities than 3. The structurally difference between them was that the latter has one more benzyloxy fragment.

In summary, we have synthesised 12 allenic aromatic ethers and identified their structures. Ten of them are new compounds, including four new natural products. Many of these allenic aromatic ethers showed potent and efficacious antitumor activities against KB and KBv200 cells. The analysis of structure–activity relationships suggested that the introduction of allenic moiety could generate or enhance cytotoxicity of these phenol compounds.

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Supplementary data

Synthetic details and characterization associated with this article are included in the Supporting Information. Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.bmcl. 2007.02.084.

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